REPORT ON SOME CARBONIC ACID TESTS ON THE WEATHERING OF MARBLES AND LIMESTONES.

By GEORGE P. MERRILL,

Head Curator, Department of Geology, United States National Museum.

The tests registered below were made with a view of determining not merely the relative solubility of certain calcareous rocks used for building and ornamental work, but as well, the manner in which the solvent acted. The ultimate aim of the experiments, as is obvious, was to ascertain how the stones would withstand the effects of an atmosphere and its rainfall made acid through absorbed carbonic acid. To make the results appreciable within a reasonable time, it was of course necessary to exaggerate the conditions. The process was as follows: Two samples of each stone selected were cut into the form of cubes approximately an inch in diameter, though without any attempt at exact correspondence in weight. How close the approximation is shown in the accompanying table of results.

The surfaces of each cube were rubbed with flour of emery on a glass plate as smooth as the nature of the material permitted, but no attempt was made to polish. They were then thoroughly washed and dried at 100° C. The cubes were then suspended by threads, in each case passed but once around the cube, in a large jar of water kept acid by a stream of carbonic acid from a charged cylinder. The water was changed once each week. No attempt was made to have the stream of bubbles constant and continuous, but the direction was changed occasionally to make certain that all were subjected to like conditions. Twice during the trial the cubes were withdrawn and while still suspended dried out by artificial heat and again immersed. At the end of 3 months they were all withdrawn, dried at a temperature of 100 degrees, and brushed off with a soft fitch brush to remove any loosened granules or dust. The appearance of each cube was carefully noted as to color changes as well as to the manner in which the solvent acted. The tables below give the weight of the cubes before and after and the loss of material both in weight and in percentage amounts. The first table gives the results of some preliminary tests which were not carried to completion, owing to imperfection of apparatus. They are, however, included here, since so far as they go they are confirmatory of those in the second. The results in both cases agree surprisingly well. It will be noted that while the amount of material lost in the first series is less than in the second, owing to the shorter period of trial, the two are always in accord. The amount of material lost by solution is not, however,

the sole item of importance, nor indeed the item of most importance. It will be noted that in some instances a stone losing a certain amount still retains a nearly smooth surface and sharp arrises. Others become roughened, granules loosened to the point of falling away, and the arrises as a consequence left ragged. In some of the stones there is a tendency for the smaller interstitial crystals to disappear, leaving the larger standing in relief. The Tennessee samples tested are of the gray and pink spotted varieties. In these the tinted calcite, which, judged from the forms, represents fragmental fossil material, is more refractory than the colorless and is left in slight relief. In the case of the oolitic limestones the oolites are eaten out, leaving the ervstalline or interstitial material and the fossil fragments in relief, the outline of the oolite being sometimes preserved by the insoluble impurities. The considerable amount of insoluble material set free from these oolitic cubes during the trial settling to the bottom of the jar as mud or remaining to be brushed off the surface when the cube was dried seems to have come wholly from the oolites. and not from the interstices. It will be noted, as might have been expected, that the dolomitic marbles were not appreciably affected and that the oolitic stones lost during the trial an amount two and three times as great as that of any other of the stones tested. In but one instance was there any marked change in color in any of the samples.

TABLE 1.

PRELIMINARY TRIAL	EXTENDING OVE	R PERIOD OF 70 DAYS.
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Kind and locality.Weight before trial in grams.Uses of the fire trial in grams.Percent- weight in grams.Percent- age loss of weight.White crystalline limestone: Marble, Pickens County, Geor- gia.44.033 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.313 44.303 44.313 44.325 46.3455 44.325 46.2345 44.7015 44.355 44.355 44.356 44.355 44.355 44.355 44.356 44.355 44.355 44.355 45.4475 44.352 44.355 45.4475 44.4765 						
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Kind and locality.	before trial in	after trial in	weight	age loss	Remarks.
Oolitic limestone: Green, Kentucky.Bowling { 41, 1655 37, 172540, 504 40, 504 37, 17250, 6115 0, 663.014 	Marble, Ynle, Colorado. White crystalline limestone: Marble, Plekens County, Geor- gia. White crystalline limestone: Marble, West Grove, Penn- sylvania. Pink crystalline limestone: Marble, Knoxville, Tennessee. Gray crystalline limestone: Marble, Concord, Tennessee. White crystalline limestone: Marble, Rutland, Vermont. Bine crystalline limestone: Marble, Rutland, Vermont. White crystalline limestone: Marble, Cartara, Italy. White crystalline dolomite: Marble, Cockeysville, Maryland White crystalline dolomite: Marble, Cockeysville, Maryland White crystalline dolomite: Marble, Tuckahoe, New York. Oolitie limestone, Bedford, Indi- ana.	$ \left\{ \begin{array}{c} 44, 313\\ 42, 3935\\ 42, 747\\ 46, 3455\\ 44, 7015\\ 54, 478\\ 45, 1075\\ 50, 4485\\ 45, 1075\\ 40, 6655\\ 41, 1245\\ 44, 414\\ 41, 0555\\ 38, 6165\\ 40, 5885\\ 38, 80\\ 36, 507\\ 42, 0655\\ 41, 405\\ 43, 5785\\ 41, 405\\ 43, 5785\\ 41, 1475\\ 5\\ 41, 1655\\ \end{array} \right. $	43, 8815 41, 925 42, 3565 46, 2345 44, 586 49, 904 47, 9725 44, 4765 40, 1185 40, 432 43, 8355 40, 432 43, 8355 40, 589 41, 3795 42, 2435 41, 3795 42, 2435 41, 3795 42, 2435 41, 3795 40, 564	$\begin{array}{c} 0, 4325\\ 0, 4585\\ 0, 3905\\ 0, 1110\\ 0, 1155\\ 0, 5045\\ 0, 6055\\ 0, 6310\\ 0, 547\\ 0, 6025*\\ 0, 6085\\ 0, 6085\\ 0, 6085\\ 0, 6085\\ 0, 6085\\ 0, 6085\\ 0, 6085\\ 0, 60925\\ 0, 0215\\ 0, 0240\\ 0, 0255\\ 1, 335\\ 1, 268\\ 0, 6115\\ \end{array}$. 0097 . 011 . 009 . 0024 . 0026 . 0108 . 0104 . 012 . 013 . 016 . 015 . 011 . 013 . 016 . 015 . 011 . 013 . 0068 . 00056 . 00056 . 00058 . 0304 . 0305 . 0414 . 0414 . 0415 . 0415 . 0414 . 0415 . 0405 . 0405. 0	f granulation. Slightly roughened; no granulation. Feffect scarcely appreciable. White portions slightly etched, leaving the pink standing in relief. White portions slightly etched, leaving the pink standing in relief. Surfaces appreciably rough- ened: no granulation. Surfaces appreciably rough- ened; like white Rutland. Not appreciably acted upon. Not appreciably acted upon. Not appreciably acted upon. Distinctly roughened and pitted, the fossil fragments left standing in relief. Distinctly roughened and pitted, the colites being enten ont, leaving sur- face covered by circular and oval pits often with a slight residual eminence in

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TABLE II.

SECOND TRIAL EXTENDING OVER PERIOD OF THREE MONTHS.

Kind of stone and locality.	Original weight in grams.	Final weight in grams.	Loss of weight in grams.	Percent- age loss in weight.	Remarks,
White crystalline limestone: Marble, Yule Creek, Colorado.	$\left\{\begin{array}{c} 51.551 \\ 48.194 \end{array}\right.$	50. 6465 47. 2465	0.9045 0.9475	0.017 .019	A very slight roughening of the surface, but no granu- lation and but slightly at- tacked on the edges or arrises.
White crystalline limestone: Marble, Pickens County, Geor- gia.	$\left. \left. \begin{array}{c} 44.034 \\ 43.6675 \end{array} \right. \right. ight.$	$\begin{array}{r} 43.315 \\ 42.946 \end{array}$	$\begin{array}{c} 0.\ 719 \\ 0.\ 7215 \end{array}$.0165 .017	Surfaces very slightly rough- cned, but no granulation.
gia. White crystalline limestone: Marble, Cherokee County, Georgia.	43.3885 44.400	$\begin{array}{c} 42.7335\\ 43.683\end{array}$	$0.655 \\ 0.717$.015 .016	Surfaces very slightly rough- ened, a slight yellowing.
White crystalline limestone: Marble, West Grove, Pennsyl- vania.	$\left\{\begin{array}{c} 47.8385\\ 47.9695 \end{array}\right.$	$\begin{array}{r} 47.\ 615\\ 47.\ 768\end{array}$	$\begin{array}{c} 0.\ 2235 \\ 0.\ 2015 \end{array}$.0047 .0042	Surfaces roughened, but no granulation.
White crystalline limestone: Marble, Proctor, Vermont. White crystalline limestone: Marble, Proctor, Vermont. White crystalline limestone: Marble, Pittsfield, Vermont.	$\left\{\begin{array}{l} 44.\ 4945\\ 44.\ 1085\\ 44.\ 974\\ 45.\ 172\\ 42.\ 536\\ 43.\ 97\end{array}\right.$	43.62 43.3125 43.9295 44.001 41.705 42.9715	0.8745 0.796 1.0445 1.171 0.831 0.9985	.019 .018 .023 .026 .019 .022	Surfaces distinctly roughened and granulated, small par- ticles loosened and falling away when handled or brushed; arrises rough- ened.
White crystalline limestone: Marble, Carrara, Italy.	$\left\{ \begin{array}{c} 43.5085\\ 42.462 \end{array} \right.$	42. 598 41. 4885	0.9105 0.9735	.021 .023	(Surfaces distinctly roughened and granulated, small par- ticles loosened and break- ing away when handled or brushed; arrises strongly attacked.
Gray crystalline limestone: Marble, Knoxville, Tennessee. Gray crystalline limestone: Marble, Knoxville, Tennessee. Gray crystalline limestone: Marble, Knoxville, Tennessee. Pink crystalline limestone: Marble, Concord, Tennessee.	$\left\{\begin{array}{l} 46.796\\ 45.1725\\ 44.069\\ 44.3095\\ 45.7165\\ 45.7565\\ 45.5565\\ 42.7345\\ 42.9635\end{array}\right.$	45. 8765 44. 2975 43. 226 43. 4545 44. 52 44. 609 41. 808 42. 1675	$\begin{array}{c} 0.9245\\ 0.875\\ 0.843\\ 0.855\\ 1.1965\\ 0.9475\\ 0.9265\\ 0.796\end{array}$.019 .019 .019 .026 .021 .021 .0185	Surfaces roughened by the corrosion of the colorless granules leaving the pink tinted standing in relief. No granulation or mechan- ical loosening of particles.
White crystalline dolomite: Marble, Cockeysville, Mary- land.	40.724 38.4355	40. 687- 38. 3995	0.037 0.036	.00091 .00093	No perceptible change.
White crystalline dolomite: Marble, Berkshire, Massachu- setts.	$ \left. \left. \begin{array}{c} 45.\ 529 \\ 45.\ 052 \end{array} \right. \right. \right. \right. \\ \left. \left. \begin{array}{c} \end{array} \right. \right. \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \left. \left. \begin{array}{c} \end{array} \right. \\ \left. \end{array} \right. \\ \left. \left. \left. \right. \\ \left. \left. \right. \right\right. \\ \left. \left. \right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \right\right. \\ \left. \left. \left. \right\right. \right\right. \\ \left. \left. \left. \left\right. \right\right. \\ \left. \left. \left. \right\right. \right\right. \\ \left. \left. \left. \left\right. \right\right. \\ \left. \left. \left. \right\right. \right\right. \\ \left. \left. \left. \left\right. \right\right. \right\right. \\ \left. \left. \left. \left\right. \right\right. \right\right. \\ \left. \left. \left. \left. \right\right. \right\right. \right\right. \\ \left. \left. \left. \left. \left\right. \right\right. \right\right. \\ \left. \left. \left. \left. \left. \right\right. \right\right. \right\right. \right\right. \right\right. \right\right. $	44. 8385 44. 378	0. 6905 0. 674	.015 .015	Surfaces distinctly roughened by corrosion along planes of cleavage and color changed to a decided buff.
White crystalline dolomite: Marble, Lee, Massachusetts.	$\left\{\begin{array}{c} 43.659\\ 44.493\\ 43.442\end{array}\right.$	43.625 44.4415 43.4025	0.034 0.0525 0.0395	.00077 .0011 .00091	No perceptible change.
White crystalline dolomite: Marble, Tuckahee, New York.	44.792 46.438 47.053	44.7465 46.3815 47.0105	0.0455 0.0575 0.0425	.0010 .0012 .0009	No perceptible change.
Oolitic limestone, Bedford, Indi- ana.	{ 36.0945 38.3245	34. 493 36. 4495	1.6015 1.875	.044 .049	Surfaces much roughened and pitted owing to solu- tion of the oolites leaving the fossil fragments and crystalline material of the interstices in relief; arrises strongly attacked.
Oolitic limestone, Bowling Green, Indiana. Oolitic limestone, Salem, Indiana	$\left\{\begin{array}{l} 38.\ 4375\\ 38.\ 6845\\ 37.\ 2795\\ 37.\ 45\end{array}\right.$	37. 1775 37. 44 35. 39 35. 591	$\begin{array}{c} 1.26 \\ 1.2445 \\ 1.8895 \\ 1.869 \end{array}$.033 .032 .0506 .050	The same, only that the stone is more distinctly oolitic and the surface becomes covered with circular and oval pits.